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RESEARCH ARTICLE

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Antimized Extrection and Formentation

Optimized Extraction and Fermentation of Antioxidant Polysaccharides from *Polygonatum odoratum*

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Abstract

Purpose: We wished to determine the optimal method for the extraction of polysaccharides with antioxidant activities from Polygonatum odoratum. Methods: Single-factor and orthogonal tests were applied to identify the best parameters for water extraction and fermentation. The solid-liquid ratio, temperature, and duration of water extraction were determined according to the total polysaccharide content. The fermentation temperature, agitation speed, inoculation amount, and fermentation time were investigated according to the optical density of Saccharomyces cerevisiae. The molecular weight distributions of water- and fermentation-extracted polysaccharides were determined by high performance gel permeation chromatography. The antioxidant activities of the polysaccharides were evaluated using 1,1-diphenyl-2-picrylhydrazyl (DPPH) and 2,2'-azino-bis(3ethylbenzothiazoline-6-sulfonic acid) ABTS radical scavenging capacities, and ferric reducing antioxidant power (FRAP). Results: The optimum water extraction parameters were 80°C for 2.5 h with a 1:60 solid-liquid ratio. For fermentation, an 8% extraction volume was inoculated with S. cerevisiae. Optimal fermentation occurred at 28°C for 28 h, with an agitation speed of 180 r/min. The molecular weights of water- and fermentation-extracted polysaccharides were 2.067×10⁴ and 9.475×10³ Da, respectively. The DPPH, ABTS, and FRAP results demonstrated an obvious increase in antioxidant activity after fermentation. Conclusion: The fermentation extraction of P. odoratum may provide valuable materials for future cosmetic use.

Keywords: *Polygonatum odoratum*; Polysaccharides; Fermentation; Molecular weight; Antioxidant activity

Introduction

Polygonatum odoratum, also called yuzhu, waisheng, weirui, or lingdangcai in Chinese, originated in the Southwest region of China and is well distributed in the wild (Lan *et al.*, 2011). Its rhizomes are widely used as an ingredient/supplement (e.g., in functional foods, flavorings, and teas). It is well known in traditional Chinese medicine (TCM) for removing dryness, promoting the secretion of fluid, and quenching thirst (Deng *et al.*, 2012; Yang *et al.*, 2010).

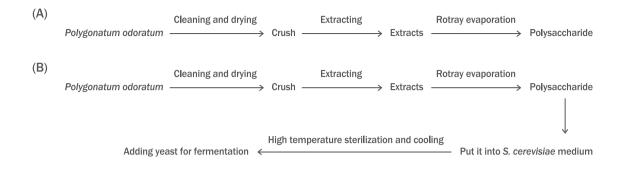
Previous phytochemical studies have shown that *P. odoratum* contains a variety of chemical constituents, such as steroidal

glycosides, dipeptides, flavonoids, and polysaccharides. *P. odoratum* polysaccharides (POPs) are of primarily functional composition, with formidable anti-tumor, antioxidant, antiinflammatory, and immunity improvement effects (Bai *et al.*, 2013; Wang *et al.*, 2009; Guo *et al.*, 2013; Qian *et al.*, 2010; Wang *et al.*, 2013). The microbial fermentation is a process of biotransformation, and has been applied for centuries in TCM. More than a thousand years ago, the Chinese people began to use microbial fermentation to enhance efficacy and potency and reduce toxicity. For example, microorganisms such as Saccharomyces cerevisiae, lactic acid bacteria, and particularly medicinal fungi like Poria cocos, Ophiocordyceps sinensis, and

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Ganoderma lucidum are used to ferment and transform TCMs, producing new prescriptions with a variety of active ingredients (Li *et al.*, 2004; Wu *et al.*, 2013).

Polysaccharides are the main active ingredients in *Polygonatum* odoratum and have good antioxidant properties (Cui, 2018). The main purpose of this article is to determine the best method for extracting antioxidant POP, in order to improve its antioxidant effect in cosmetics and better develop the efficacy of POPs.

Methods

1. Chemicals and reagents

P. odoratum was collected in Hunan province, China. *S. cerevisiae* (109 colony-forming units [CFU]/mL) was purchased from China General Microbiological Culture Collection Center. Ethylbenzothiazoline-6-sulphonic acid (ABTS; Sigma-Aldrich, USA), 1,1-diphenyl-2-picrylhydrazyl (DPPH; Tokyo Chemical Industry, Japan), potassium peroxydisulfate (Xilong Chemical Co., Ltd., China), TPTz (Adamas Reagent Co., Ltd., China), iron chloride hexahydrate (Shanghai Macklin Biochemical Co., Ltd., China), iron (II) sulfate heptahydrate (Xilong Chemical Co., Ltd., China), hydrochloric acid (Beijing Chemical Works, China), and L-ascorbic acid (J&K Scientific Co., Ltd., China) were all of analytical grade.

2. Extraction methodds

Water extraction: The medicinal material of *Polygonatum* odoratum is washed, dried in a drying oven (Rotary evaporator; Shanghai Ailang Instrument Co., LTD; China), and crushed by a crusher (Rocking crusher; Guangzhou Xulang machinery Equipment Co., LTD; China) to obtain the medicinal powder.

Weigh 10 g powder and add 300mL deionized water at 1:30 (m/V), and extract it twice at 100°C by hot reflux for 120 min each time. The extract was mixed and filtered through a 0.45 μ m filter plate to obtain the polysaccharide extract of *Polygonatum* odoratum. The extract was concentrated by Rotary evaporator evaporation at 65°C. The flowchart is shown in Figure 1A.

Fermentation process: Polysaccharides were extracted from dried samples using the optimal parameters. The supernatant was collected in a 250 mL vacuum flask with *S. cerevisiae* medium. The clear liquid was autoclaved at 121°C for 20 min for sterilization. When the liquid temperature decreased to 30°C, activated *S. cerevisiae* was inoculated into the liquid and fermentation was initiated in a shaker (Jia *et al.*, 2006). The flowchart is shown in Figure 1B.

Fermentation will be carried out on the polysaccharide extract obtained after optimizing the water extraction conditions. The fermentation method will produce various secondary metabolites, which will significantly change the drug properties and improve its efficacy. By optimizing two extraction methods and comparing the antioxidant effects of polysaccharides obtained, a more suitable extraction method for POPs was determined.

3. Total polysaccharide content

The total polysaccharide content was determined using the improved phenol-sulfuric acid method (Liu *et al.*, 2009). Measurements were acquired at 490 nm using a microplate reader (SpectraMax 190; Molecular Devices LLC). The total polysaccharide content was calculated as glucose/g using a calibration curve, Samples were analyzed in triplicate.

4. Single-factor tests for optimal water extraction

To obtain the appropriate conditions for maximum

polysaccharide extraction, temperatures of 60, 70, 80, 90, and 100°C, liquid-solid ratios of 1:20, 1:30, 1:40, 1:50, and 1:60, and durations of 1.0, 1.5, 2.0, 2.5, and 3.0 h were independently tested. The total polysaccharide content was determined by the phenol-sulfuric acid method.

5. Orthogonal test for water extraction

According to the results of the single factor experiment, an L9 (34) orthogonal array with three factors at three levels was used to identify optimum conditions for the maximal extracted polysaccharide yield (Choi & Park, 2002; Wang *et al.*, 2015). The used factors and the levels of orthogonal design are shown in Table 1.

6. Fermentation process

Polysaccharides were extracted from dried samples using the optimal parameters. The supernatant was collected in a 250 mL vacuum flask with *S*, *cerevisiae* medium. The clear liquid was autoclaved at 121 °C for 20 min for sterilization. When the liquid temperature decreased to 30 °C, activated *S*, *cerevisiae* was inoculated into the liquid and fermentation was initiated in a shaker (Jia *et al.*, 2006; Zhu *et al.*, 2012).

7. Single-factor tests for optimal fermentation

To determine the appropriate conditions for maximum *S. cerevisiae* growth, inoculation volumes of 2, 4, 6, 8, and 10%; rotary speeds of 100, 120, 140, 160, and 180 r/min; temperatures of 26, 28, 30, 32, 34° C; and the effects of time on the yield of *S. cerevisiae* were independently tested. The optical density of *S. cerevisiae* was determined at 600 nm by microplate reader.

8. Orthogonal test for fermentation optimization

Table 1. Orthogonal test for water extraction

According to the results of the single factor experiments, an L9 (34) orthogonal array with four factors at three levels was used to determine the optimal conditions for maximal *S. cerevisiae* yield. The factors used and the levels of orthogonal design are shown in Table 2.

9. Molecular weight distribution assay

The molecular weights of water-and fermentation-extracted polysaccharides were determined using a Waters e2695 Alliance high performance gel permeation chromatography system equipped with a DAWN HELEOS-II Multi-Angle static Light Scattering detector (Wyatt Technologies) and a refractive index detector. All samples (3.0 mg) were dissolved in 0.1 M NaNO₃, passed through a 0.45 μ m filter, and applied to a Shodex SUGAR KS-805/KS-803 gel-filtration chromatographic column. The column was maintained at 60°C and eluted with 0.1M NaNO₃ at a flow rate of 0.8 mL/min.

10. Determination of DPPH radical scavenging activity

DPPH radical scavenging rates in the extracts were measured according to Brand *et al.* (1995). Samples (1.0, 2.0, 4.0, 6.0, 8.0, and 10.0 mg/mL) were added to 1.0 mL of 0.2 mM ethanolic DPPH and shaken. After 30 min, the decrease in absorbance was measured at 517 nm. Methanol was used as a blank, and ascorbic acid was used as a positive control. Assays were performed in triplicate.

11. Scavenging of ABTS radicals

The radical scavenging activities of the extracts against ABTS radical cations were measured using the method of Li *et al.* (2012) with some modifications. ABTS was dissolved in water to 7 mmol/L. ABTS radical cations were produced by reacting

No.	Temperature (°C)	Liquid-solid ratio	Time (h)
1	90	1:40	2.5
2	80	1:60	2.0
3	100	1:50	1.5

Table 2. Orthogonal test for fermentation parameters

No.	Inoculation volume (%)	Rotary speed (r/min)	Temperature (°C)	Time (h)
1	4%	140	28	20
2	6%	160	30	24
3	8%	180	32	28

an ABTS stock solution with 2.45 mmol/L potassium persulfate and allowing the mixture to stand in the dark at 25°C for 12– 16 h before use. The ABTS radical cation solution was diluted in ethanol to an absorbance of 0.70 ± 0.02 at 734 nm and equilibrated at 30°C. Samples (0.1 mL; final concentrations of 0.1, 0.5, 1.0, 2.0, 4.0, and 6.0 mg/mL) were mixed with 3.9 mL of diluted ABTS radical cation solution, and the absorbance at 734 nm was measured after reaction for 6 min.

12. Total antioxidant capacity assay

The total antioxidant capacity of the extracts was measured by the ferric reducing antioxidant power (FRAP) method (Wang *et al.*, 2021; Nilsson *et al.*, 2005). We added 180 µL FRAP working liquid into each well of a 96–well plate, then added 5.0 µL of FeSO₄ standard solution at concentrations of 0.15, 0.3, 0.6, 0.9, 1.2, and 1.5 mM to the standard curve detection wells. Sample wells received 5.0 µL of the extracted samples in triplicate, and a 0.15 mM sample containing 1.5 mM Trolox was used as a positive control. The plate was gently shaken and incubated at 37° for 5–7 minutes, and then the decrease in absorbance was measured at 593 nm. Distilled water was used as a blank. The total antioxidant capacity of the sample was calculated according to the standard curve.

13. Statistical analyses

Data analyses were performed using IBM SPSS Statistics version 22.0 (IBM Corporation, USA). The data was performed by Probit regression analysis.

Results

1. Optimization of extraction conditions for water-soluble polysaccharides

Single-factor and orthogonal tests were used to determine the optimal extraction conditions for water-soluble polysaccharides. The effects of extraction temperature, ratio of raw material to water, and extraction time on the polysaccharide yield are shown in Figure 2. The optimal temperature was 90°C, and the optimal liquid-solid ratio was 1:50. Considering energy and efficiency, the optimal extraction time was 2.0 h.

An L9 (33) orthogonal array was used to optimize the extraction parameters, including nine experiments corresponding to nine rows and three columns. The results of all tests are

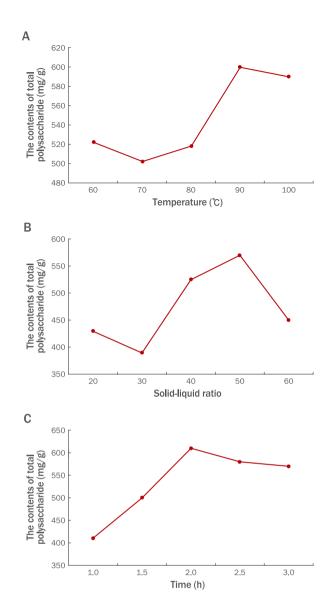


Figure 2. The contents of total polysaccharide of single-factor test for water extraction.

(A) Temperature; (B) Solid-liquid ratio, (C) Extraction time. The solid-liquid ratio and extraction time remain unchanged by changing the extraction temperature: 60, 70, 80, 90, 100 °C; The extraction temperature and extraction time were unchanged by changing the solid-liquid ratio: 20, 30, 40, 50, 60; The extraction temperature and solid-liquid ratio were unchanged by changing the extraction time: 1.0, 1.5, 2.0, 2.5, 3.0 h, and the improved phenol-sulfuric acid method was used to test the polysaccharide content in the extract, so as to determine the optimal extraction temperature, extraction time and solid-liquid ratio.Descriptive statistical analysis was performed on the data .

shown in Table 3. The F-value was used to qualitatively identify effective factors. The liquid-solid ratio was of highest importance for effective polysaccharide extraction, followed by temperature and extraction time. The optimum extraction parameters were

No.	Temperature (°C)	Liquid-solid ratio	Time (h)	Content of total polysaccharide (mg/g)
1	1	1	1	680.6
2	1	2	2	759.4
3	1	3	3	566.4
4	2	1	2	646.2
5	2	2	3	763.0
6	2	3	1	652.7
7	3	1	3	576.4
8	3	2	1	668.6
9	3	3	2	589.8
k ₁	668.8	634.4	667.3	
k ₂	687.3	730.3	665.2	
k ₃	611.6	603.0	635.2	
R	75.7	127.4	32.1	

Table 3. Results and analysis of orthogonal test

k values indicate the best extraction parameters; R values shows the influence of the three factors decreased in the following order: Liquid-solid ratio>Temperature>Time.

Table 4	 Results and 	analysis of	orthogonal test
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No.	Inoculation volume (%)	Rotary speed (r/min)	Temperature (°C)	Time (h)	A-600nm	
1	1	1	1	1	1.021	
2	1	2	2	2	0.962	
3	1	3	3	3	1.072	
4	2	1	2	3	1.042	
5	2	2	3	1	0.907	
6	2	3	1	2	1.021	
7	3	1	3	2	1.111	
8	3	2	1	3	1.099	
9	3	3	2	1	1.112	
k1	1.018	1.058	1.047	1.013		
k_2	0.990	0.989	1.039	1.032		
k ₃	1.107	1.068	1.030	1.071		
R	0.117	0.079	0.016	0.058		

k values indicate the best extraction parameters; R values shows the influence of the three factors decreased in the following order: Inoculation volume>Rotary speed>Time>Temperature.

a 1:60 liquid-solid ratio, an 80°C extraction temperature, and a 2.5 h extraction time. Under optimal conditions, the total polysaccharide contents were 83.49%.

2. Optimization of fermentation conditions

Single-factor and orthogonal tests were applied to determine the appropriate condition for maximum S, *cerevisiae* growth. The optimal fermentation time was 24 h. Effects of inoculation volume, rotary speed, and temperature on the fermentation process are shown in Figure 3, and the growth phase of S. *cerevisiae* is shown in Figure 3D. The optimal inoculation volume was 6%, and the optimal rotary speed was 160 r/min. The most suitable temperature for fermentation was 28°C.

An L9 (34) orthogonal array was used to further optimize the fermentation conditions. The results are shown in Table 4. The best fermentation conditions were an inoculation volume of 8%, a rotary speed of 180 r/min, a 28 h fermentation time, and a temperature of 28°C. The inoculation volume was of highest importance for effective fermentation, followed by rotary speed, fermentation time, and temperature.

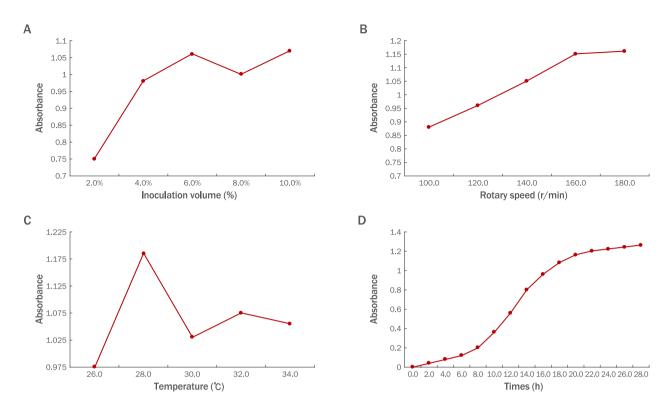


Figure 3. Single-factor test for fermentation parameters on the growth of S.cerevisiae.

(A) Inoculation amount; (B) rotary speed; (C) temperature; (D) time. Univariate and orthogonal experiments were used to determine the optimal incubation temperature, time, rotation speed and inoculum amount for culturing Saccharomyces cerevisiae, and descriptive statistical analysis was performed on the data.

3. Molecular weight distribution

The molecular weight distributions of water and fermentation– extracted polysaccharides were 2.067×10^4 and 9.475×10^3 Da, respectively, indicating that the molecular weight after fermentation extraction was lower and more homogeneous than that of water–extracted polysaccharides.

4. DPPH radical scavenging activity

The antioxidant capacities of water- and fermentationextracted polysaccharides were examined by the DPPH method. Polysaccharides extracted by water and fermentation exhibited DPPH scavenging activity with concentrations showing 50% inhibition (IC₅₀ values) of 4.9 and 1.0 mg/mL, respectively. Fermentation produced better DPPH radical scavenging activity than water extraction (Figure 4A).

5. ABTS radical scavenging activity

ABTS is converted to its radical cation (ABTS \cdot +) by reacting with a strong oxidizing agent (e.g., potassium permanganate,

potassium persulfate). This radical cation is blue and absorbs light at 734 nm, and can be converted back to its colorless neutral form by hydrogen-donating antioxidants. The ABTS+ inhibitive efficiencies of water- and fermentationextracted polysaccharides are presented in Figure 4B. Waterand fermentation-extracted polysaccharides showed ABTS radical scavenging activity, with IC₅₀ values of 1.4 and 0.3 mg/ mL, respectively. Fermentation produced better ABTS radical scavenging activity than water extraction.

6. Total antioxidant capacity

The total antioxidant capacities of water- and fermentationextracted polysaccharides (0 to 10 mg/mL) were examined by FRAP. The standard curve provided the equation y=0.411x-0.0755 (R²=0.9946, x: concentration of FeSO₄ (mM); y: absorbance). The total antioxidant capacity of each sample was expressed as the FeSO₄ concentration. Fermentation produced better total antioxidant capacity than water extraction (Figure 4C, 4D).

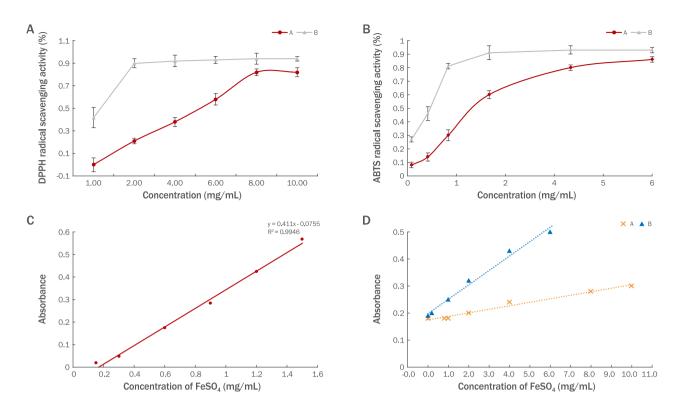


Figure 4. Radical scavenging activity and total antioxidant capacity of water- and fermentation-extracted polysaccharides. (A) DPPH free radical scavenging assay was used to study the antioxidant effect of polysaccharides under different conditions concentrations: 1, 2, 4, 6, 8, and 10 mg/mL. Data are presented as mean±SD. (B) ABTS radical scavenging assays were conducted to investigate the antioxidant effect of polysaccharides under different conditions concentrations: 0.1, 0.5, 1, 2, 4 and 6 mg/mL. Data are presented as mean±SD. (C) Standard curve of total antioxidant power. (D) Total antioxidant capacity. Sample A: water polysaccharides; Sample B: fermentation-extracted polysaccharide (n=3).The total antioxidant capacities of water- and fermentation-extracted polysaccharides (0 to 10 mg/mL) were examined by FRAP.

Discussion

Maceration, heat reflux, ultrasound assistance, and acidic hydrolysis are common methods for extracting polysaccharides from plant tissues, but they require long extraction times, high extraction temperatures, or expensive equipment. In addition, they may pollute the environment (Qian, 2014). Fermentation is one of extraction method highly influenced by various parameters including the nature of the substrate, pH of the medium, nutrient availability, inducer supplementation, fermentation temperature, etc. (Singhania *et al.*, 2010). Compared with traditional solvent extraction method, fermented small molecule has advantage in skin absorption, and the fermentation process is environmental friendly.

P. odoratum is rich in dietary fibers, whose functional properties are closely related to some of its therapeutic effects, such as reducing the risk of coronary heart disease, diabetes, obesity, and some forms of cancer, and lowering cholesterol

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and fat levels (Liu *et al.*, 2015; Lan *et al.*, 2012). Most dietary fibers are components of plant cell wall polysaccharides, which are resistant to digestion by the alimentary enzymes of humans (Bhat, 2000). *S. cerevisiae* is commonly used to ferment materials. Fermented foods or traditional medicines are the result of extensive microbial growth, during which microbial metabolism and biosynthesis pathways transform the starting materials to products with distinct organoleptic properties (Wolfe & Dutton, 2015).

In the present study, the antioxidant activities of POPs were obviously increased by fermentation. Molecular weight distribution analysis indicated that polysaccharide hydrolysis during fermentation increased antioxidant activities. Therefore, fermentation is a preferred technique to extract antioxidant POPs. Subsequent research can apply the fermented polysaccharides extracted from *Polygonatum odoratum* to cosmetics with anti-aging effects, and use human experiments or mouse skin model experiments to test their antioxidant effects, further demonstrating their efficacy in products.

Conclusion

In this study, we developed an effective fermentationbased extraction method for POPs using single-factor and orthogonal tests. The optimal fermentation method used an extraction inoculation volume of 8%, rotary speed of 180 r/min, fermentation time of 28 h, and fermentation temperature of 28°C. During fermentation, the decreased molecular weight of the POPs produced was consistent with increased antioxidant activity. Our results suggest the potential use of fermented POPs as a material in foods, medicines, or cosmetics.

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Author's contribution

LL and MG designed this study. JL conducted research. ZG and HH provided help and suggestions for this study. JL analyzed these data and wrote the manuscript. All authors contributed to the editorial changes in the manuscript. All authors read and approved the final manuscript.

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국문초록

Polygonatum odoratum에서 항산화 다당류의 추출 및 발효를 위한 최적화

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목적: 본 연구는 Polygonatum odoratum으로부터 항산화 활성을 갖는 다당류를 추출하기 위한 최적의 방법을 결정하고자 한다. 방 법: 최적의 물 추출 발효 공정 변수를 결정하기 위해 단일 요인 및 직교 실험을 진행하였다. 총 다당류 함량에 따라 물질 대 액체 비 율, 온도, 물담근 시간을 결정하였다. Saccharomyces cerevisiae의 흡광도를 지표로 하여 발효온도, 교반속도, 접종량, 발효시간을 조사하였다. 물 추출 다당류와 발효 추출 다당류의 분자량 분포를 결정하기 위해 고성능 겔 투과 크로마토그래피를 사용하였다. 다 당류의 항산화 활성은 DPPH와 ABIS 라디칼 소거능, FRAP (철환원 항산화력)을 이용하여 평가하였다. 결과: 최적의 물 추출 매 개변수는 고액비1:60, 온도 80℃, 추출시간 2.5 h이다. 발효를 위해 추출 부피의 8%에 S. cerevisiae를 접종하였다. 최적의 발효조 건은 28℃에서 28 h, 교반속도는 180 r/min이다. 물추출 다당류와 발효추출 다당류의 분자량은 각각 2.067×10⁴ Da, 9.475×10³ Da이었다. DPPH, ABIS, FRAP 항산화 활성이 발효 후 크게 항상되었다. 결론: Polygonatum odoratum의 발효 및 추출 과정은 Polygonatum odoratum의 화장품산업에 응용될 수 있는 원료를 제공할 수 있다고 사료된다.

핵심어: Polygonatum odoratum, 당류, 발효, 분자량, 항산화

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中文摘要

玉竹抗氧化多糖的优化提取与发酵

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目的:研究从玉竹中提取抗氧化多糖的最佳工艺。方法:采用单因素试验和正交试验确定最佳水提发酵工艺参数。根据多糖总含量确定料液比、温度和水浸时间。以酿酒酵母的光密度为指标,考察发酵温度、搅拌速度、接种量和发酵时间。采用高效凝胶渗透色谱法测定了水提多糖和发酵提多糖的分子量分布。以DPPH和ABTS自由基清除能力和铁还原抗氧化能力(FRAP)评价多糖的抗氧化活性。结果:最佳水提工艺参数为80℃,料液比1:60,提取2.5 h。发酵时,以8%的萃取量接种酿酒酵母。最佳发酵条件为28℃发酵28 h,搅拌速度为180 r/min。水提多糖和发酵多糖的分子量分别为2.067×10⁴和9.475×10³ Da。发酵后DPPH、ABTS和FRAP的抗氧化活性明显提高。结论:玉竹发酵提取工艺可为玉竹的美容应用提供有价值的原料。

关键词: 玉竹,多糖,发酵,分子量,抗氧化活性